M1. (a) protons, electrons both required, either order 1 neutrons 1 electron, nucleus both required, this order 1 2.7 (days) (b) allow **1** mark for showing correct use of the graph 2 (C) put source into water at one point on bank accept the idea of testing different parts of the river bank at different times 1 see if radiation is detected in polluted area accept idea of tracing or put source into water at three points on bank (1) see if radiation is detected downstream of factory or farmland or sewage treatment works (1)

M2. (a) 146
(b) atomic number
(c) (i) alpha
(ii) number of protons changes accept atomic number changes

accept <u>loses or gains</u> protons do **not** accept protons with any other particle e.g. number of protons and neutrons changes incorrect do **not** accept any reference to mass number 1

1

1

M3. (a) (i) (atoms / elements with) the same number of protons but different numbers of neutrons accept (atoms / elements with) different mass number but same atomic number

> substances that give out radiation accept alpha, beta or gamma for radiation accept an unstable nucleus that decays radioactive decay takes place is insufficient

(b) 85 years

± 2 years allow **1** mark for showing correct method on the graph

2

1

1

(c) (i) a helium nucleus accept 2 neutrons and 2 protons accept 2⁴He do **not** accept helium atom

1

1

1

1

 the rate of decay (of plutonium) decreases accept fewer (plutonium) nuclei (to decay) accept radioactivity decreases

less heat produced do **not** accept energy for heat

(d) (i) (outside the body)

alpha (particles) cannot penetrate into the body

(inside the body)

(heat produced from decay) damages / kills cells / tissues

accept causes cancer for damages / kills cells / tissues accept **highly** toxic

(ii) any **one** from:

- worried same could happen again
- an accident may cause radiation to be spread around the Earth / atmosphere
- idea of soil contamination resulting from accident / release of radioactive material
- idea of negative effect on health resulting from accident / release of radioactive material accept any sensible suggestion

1

M4. (a)	(i)	(total) number of protons plus neutrons accept number of nucleons accept amount for number do not accept number of particles in the nucleus	1
	(ii)	number of neutrons decreases by one	1
		number of protons increases by one accept for both marks a neutron changes into a proton	1
(b)	(i)	²⁰⁸ Th 81	1
		correct order only	1
	(ii)	the number of protons determines the element accept atomic number for number of protons	1
		alpha and beta decay produce different changes to the number of pro there must be a comparison between alpha and beta which is more than a description of alpha and beta decay alone oralpha and beta decay produce different atomic numbers ignore correct reference to mass number	otons 1

[7]

M5 . (a)	BEG	all 3 required and no other any order	1
		ber of / 00 protone (and different numbers of neutrone)	1
	same num	ber of / 88 protons (and different numbers of neutrons) same number of electrons is insufficient	1
(b)	(i) 222		1
	86		1
	(ii) 4800) allow 1 mark for obtaining 3 half-lives	2
(c)	ethical		1
	or value own	lied to (about safety of working conditions) accept (women) not warned of the dangers given no protection is insufficient / scientists' lives more than women or at women humanely	1
(d)	eg too many i	y sensible suggestion interests in continued use of radium may cause public unrest do not accept not enough evidence	

doctors not want to be blamed for illnesses (caused by radium) accept doctors not wanting to be sued (for harm caused by using radium)

doctors thought (possible) benefits outweighed (possible) risks do **not** accept did not know radium could be harmful believe radium could treat illnesses is insufficient

M6. (a) has an equal amount of positive charge accept pudding/it is positive

(b) (experimental) results could not be explained using 'plum pudding' model**or**(experimental) results did not support plum pudding model accept (experimental) results disproved plum pudding model

(c) (i) **A** – most of atom is empty space**or**most of atom concentrated at the centre

B – nucleus is positive (so repels alpha particles)
 accept nucleus has the same charge as alpha

C – nucleus is very small accept nucleus is positive if not scored for B

ornucleus is a concentrated mass accept nucleus has a very concentrated charge

 (ii) (if predictions correct, this) supports the new model answers should be in terms of the nuclear model accept supports his/new/nuclear theory accept proves for supports accept shows predictions/ Rutherford was correct

[6]

1

1

1

1

1

1

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M7.any two pairs from:

to gain credit it must be clear which model is being described do **not** accept simple descriptions of the diagram without comparison

nuclear model mass is concentrated at the centre / nucleus (1)
 accept the nuclear model has a nucleus / the plum pudding model does not have a nucleus for 1 mark

plum pudding model mass is evenly distributed (1)

• nuclear model positive charge occupies only a small part of the atom (1)

plum pudding model positive charge spread throughout the atom (1)

nuclear model electrons orbit some distance from the centre (1)
 accept electrons in shells / orbits provided a valid
 comparison is made with the plum pudding model

plum pudding electrons embedded in the (mass) of positive (charge) (1) do **not** accept electrons at edge of plum pudding

• nuclear model the atom mainly empty space (1)

plum pudding model is a 'solid' mass (1)

[4]

M8.	(a)	cannot predict <u>which</u> dice / atom will 'decay' accept answers given in terms of 'roll a 6'	1
		cannot predict <u>when</u> a dice / atom will 'decay'	1
	(b)	3.6 to 3.7 (rolls) allow 1 mark for attempt to read graph when number of dice = 50	2
	(c)	90	1
	(d)	uranium	1
	(e)	beta	1
		proton number has gone up (as neutron decays to proton and e^-)	1
	(f)	prevents contamination or prevents transfer of radioactive material to teacher's hands	1
		which would cause damage / irradiation over a longer time period.	1 [10]